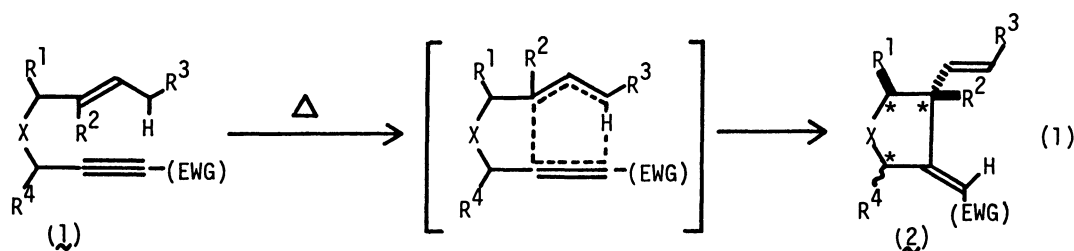


Stereochemical Control in Intramolecular Ene Reactions  
of Crotyl Propargyl Ether Systems

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Stereochemical features are described of the title reactions leading to the  $\gamma$ -alkylidene- $\beta$ -vinyloxolanes. Of particular interest is the high 1,2-trans diastereoselectivity observed for the ( $\alpha$ -alkyl)crotyl and ( $\alpha,\beta$ -dialkyl)crotyl ether systems with methoxycarbonyl group at the acetylenic terminus which remarkably facilitates the ene cyclization.

Recently the intramolecular ene reaction has been emerging as one of efficient tools for (carbo)cyclizations.<sup>1,2)</sup> However, the stereochemistry of this type of pericyclic reaction has not been fully explored, compared with mechanistically-related intramolecular Diels-Alder reactions.<sup>2)</sup> As part of a program designed to develop the intramolecular ene reaction into a new and stereoselective methodology for five-membered cyclization, we have been investigating the stereochemistry of the ene cyclization as formulated in Eq. 1, where an acetylenic bond acts as an internal enophile.<sup>3,4)</sup> Our major concerns are associated with stereocontrol over the newly created chiral centers. Disclosed herein are the stereochemical observations on the thermal ene cyclization of propargyl ether systems 1 [X=O] with and without methoxycarbonyl group at the acetylenic terminus.

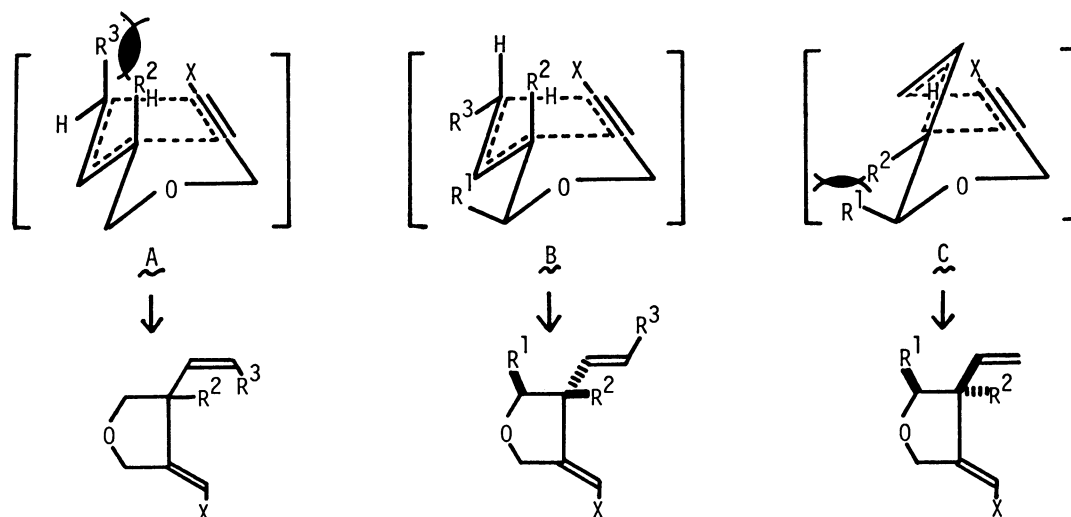


Substrate ethers (1) with a variety of substitution patterns were prepared by the standard methods<sup>5)</sup> and heated in toluene under argon atmosphere in sealed tubes. Table I summarizes the products and their stereochemistries, which reveal the following significant features of the present ene cyclization.

(a) The ene cyclization is remarkably facilitated by the introduction of methoxycarbonyl group at the acetylenic terminus (e.g., entry 1 vs. 2), as previously reported for related ene reactions.<sup>3c)</sup> (b) Entry 3 is of special

interest because the formation of  $2c$  as a sole product indicates that the ene cyclization involving the acetylenic enophile prevails over the counterpart involving the olefinic enophile.<sup>6)</sup> (c) The ene cyclization results in the exclusive formation of (Z)-geometry with respect to the newly formed alkylidene part<sup>7)</sup> (e.g., entries 2 and 3). (d) With respect to the newly created vinylic part, on the other hand, relatively high levels of (E)-selection was observed (entries 4 and 5). Of particular interest is the remarkably enhanced (E)-selectivity observed in entry 5. (e) The ene cyclization provides only low levels of 1,3-asymmetric induction (entries 6 and 7). (f) In contrast, relatively high levels of 1,2-asymmetric induction, i.e., trans diastereoselection,<sup>8)</sup> were observed (entries 8-14). Particularly noteworthy is the remarkably high diastereoselectivity observed with the substrates bearing the bulky  $\alpha$ -substituents ( $R^1$ ) (entries 11-14). It should be noted here that quaternary centers can be generated stereoselectively as observed with the ( $\beta$ -methyl)crotyl ether systems (entries 13 and 14).

Among these stereochemical observations, the trends (d) and (f) are of synthetic value and of mechanistic interest. The two trends are reasonably explicable in terms of the possible transition states (A, B, and C) as follows.



Of the two transition states A and B, the latter leading to the (E)-vinylic part is sterically more favorable since the former would suffer a large pseudo-1,3-diaxial repulsion of  $R^2 \leftrightarrow R^3$ . The larger 1,3-repulsion in A ( $R^2 = CH_3$ ) than that in B ( $R^2 = H$ ) accounts for the enhanced (E)-selectivity observed for  $1e$ . Of the two transition states B and C, on the other hand, the endo transition state (B) leading to the trans-configuration is sterically less congested because the exo transition state (C) would suffer a gauche interaction of  $R^1 \leftrightarrow R^2$ . The larger gauche repulsion in C ( $R^1 = i\text{-Pr}$  or  $Ph$ ) than that in B ( $R^1 = CH_3$ ) is responsible for the enhanced trans-selectivity observed for  $1k$ - $1n$ . Thus, we believe that the bicyclic transition-state model might provide a guiding principle for explaining and predicting the stereochemistry of the present type of five-membered ene cyclization in general.

Table 1. Thermal Ene Cyclization of Ethers 1

Run	Substrate	Temp/°C	Time/h	Product <sup>a)</sup>	
				Yield/% <sup>b)</sup>	Stereochemistry
1	<u>1a</u> , X=H	250	6	<u>2a</u> 95	
2	<u>1b</u> , X=CO <sub>2</sub> Me	200	1	<u>2b</u> 98	
3	<u>1c</u> , X=CH <sub>3</sub>	250	17	<u>2c</u> 41	Only <u>z</u> <sup>c,d)</sup>
4	<u>1d</u> , R=H	200	14	<u>2d</u> 93	87 : 13 <sup>e,f)</sup>
5	<u>1e</u> , R=CH <sub>3</sub>	200	48	<u>2e</u> 58	>95 : <5 <sup>e,f)</sup>
6	<u>1f</u> , X=H	200	52	<u>2f</u> 83	60 : 40 <sup>c,g)</sup>
7	<u>1g</u> , X=CO <sub>2</sub> Me	170	24	<u>2g</u> 77	60 : 40 <sup>c,g)</sup>
8	<u>1h</u> , X=H, R=CH <sub>3</sub>	200	5.5	<u>2h</u> 73	76 : 24 <sup>c,h)</sup>
9	<u>1i</u> , X=SiMe <sub>3</sub> , R=CH <sub>3</sub>	200	4.5	<u>2i</u> 71	81 : 19 <sup>c,h)</sup>
10	<u>1j</u> , X=CO <sub>2</sub> Me, R=CH <sub>3</sub>	130	24	<u>2j</u> 90	86 : 14 <sup>c,h)</sup>
11	<u>1k</u> , X=CO <sub>2</sub> Me, R= <u>i</u> -Pr	130	24	<u>2k</u> 90	>95 : <5 <sup>h,i)</sup>
12	<u>1l</u> , X=CO <sub>2</sub> Me, R=Ph	130	24	<u>2l</u> 97	93 : 7 <sup>h,i)</sup>
13	<u>1m</u> , R= <u>i</u> -Pr	130	24	<u>2m</u> 93	90 : 10 <sup>h,i)</sup>
14	<u>1n</u> , R=Ph	130	24	<u>2n</u> 88	92 : 8 <sup>h,i)</sup>

a) All the products were characterized by IR and <sup>1</sup>H and <sup>13</sup>C NMR. b) Isolated yield after column chromatography. c) Determined by <sup>1</sup>H NMR. d) The (Z)-geometry of the alkyldiene part was proved by the single peak due to the methyl which appeared as a triple doublet (J=1.2 and 7.8 Hz). e) Determined by capillary GLC (XE-60, 30 m). f) Refers to the E/Z ratio for the vinylic part. The configuration of the major isomer (separated by preparative GLC) was assigned to E on the basis of IR absorption at 965 cm<sup>-1</sup>. g) Stands for the diastereomeric ratio. No attempt was made to assign the stereochemistry. h) Refers to the trans/cis ratio. For the stereochemical assignments, see Ref. 8. i) Determined by HPLC (Zorbax SIL, hexane/ethyl acetate).

In summary, this work has now revealed the stereochemical features of the thermal ene cyclization leading to the five-membered rings. On the basis of these observations, we are currently investigating applications of this methodology to natural product synthesis.

## References

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- 8) The trans-configuration of the major diastereomer was assigned through comparison of its <sup>13</sup>C NMR spectrum with that of the minor diastereomer. The configuration of the isomer of which C<sub>α</sub> resonates at lower fields was assigned to trans on the basis of the <sup>13</sup>C NMR trends reported for closely-related oxolane derivatives: E. L. Eliel, V. S. Rao, and K. M. Dietrusiewicz, *Org. Magn. Reson.*, **12**, 461 (1979). The δ-values of C<sub>α</sub> for the trans and cis isomers are as follow; 2h: 80.19, 78.35; 2i: 79.48, 77.48; 2j: 78.94, 77.05; 2k: 87.61, 87.39; 2l: 84.57, 82.73.  
The trans-configuration of the major diastereomer of 1m was ascertained by 500 MHz 2-D NOE spectroscopy (NOESY), where a cross peak between the α-methine proton and the vinylic proton was observed. In contrast, the cross peak was not observed in the NOESY of the minor cis-1m.  
The major diastereomer of 1n of which the β-methyl <sup>1</sup>H NMR signal appears at a remarkably higher field was assigned to the trans isomer on the basis of a ring-current effect of the benzene nucleus.

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